Hybridization of Complementary and Homologous Peptide Nucleic Acid Oligomers to a Guanine Quadruplex-Forming RNA[†]

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ABSTRACT: Peptide nucleic acid (PNA) oligomers targeted to guanine quadruplex-forming RNAs can be designed in two different ways. First, complementary cytosine-rich PNAs can hybridize by the formation of Watson—Crick base pairs, resulting in hybrid PNA—RNA duplexes. Second, guanine-rich homologous PNAs can hybridize by the formation of G tetrads, resulting in hybrid PNA—RNA quadruplexes. UV thermal denaturation, circular dichroism, and fluorescence spectroscopy experiments were used to compare these two recognition modes and revealed 1:1 duplex formation for the complementary PNA and 2:1 (PNA2—RNA) quadruplex formation for the homologous PNA. Both hybrids were very stable, and hybridization was observed at low nanomolar concentrations. Hybrid quadruplex formation was equally efficient regardless of the PNA strand polarity, indicating a lack of interaction between the loop nucleobases on the PNA and RNA strands. The implications of this finding on sequence specificity as well as methods to improve affinity are also discussed.

The guanine (G) quadruplex structure arises from the simultaneous hydrogen bonding of four guanine nucleobases in a quasi-planar array, with multiple such G quartets associated cofacially via π stacking (1-3). Binding of specific metal ions within the G-tetrad stack stabilizes the overall structure (4, 5). G quadruplexes are formed by oligonucleotides based on telomeric repeat sequences (5) and have been implicated in several biological processes (6), including regulating transcription (7-9), mediating RNA—protein interactions (10-14), and controlling alternative RNA splicing (15, 16) and translation (17). They are also commonly selected from combinatorial synthetic DNA or RNA libraries screened for binding to a variety of protein targets (12, 18-23).

The growing awareness of the biological importance of G quadruplexes has motivated considerable interest in developing synthetic compounds that can bind selectively to these structures (24). A wide variety of G-quadruplex-binding ligands have been discovered, and these exhibit varying degrees of affinity and selectivity, with selectivity relating both to quadruplex versus duplex nucleic acid structures as well as between different quadruplex topologies (25, 26). Thus, these compounds are intended to recognize specific nucleic acid structures.

Our research focuses on targeting G-quadruplex-forming sequences, rather than G-quadruplex structures. In one approach, a Watson—Crick complementary oligomer is synthesized and targeted to some subsequence within the overall G-quadruplex target. Because of the high percentage of guanine in the target, a large number of G—C pairs are formed upon hybridization of the complementary oligomer.

Combining a high G-C content with a high-affinity DNA mimic such as peptide nucleic acid (PNA) allows very short complementary oligomers to bind and form stable hybrid duplexes. This strategy has been shown to be effective for both DNA (27, 28) and RNA (29) quadruplex targets.

An alternative approach to recognition of quadruplex-forming sequences involves synthesizing *homologous* oligomers, in which case both the target and the probe have the same G-rich sequence of nucleobases. Homologous PNAs form hybrid G quadruplexes with both DNA (30) and RNA targets (29). These structures exhibit stabilization by specific metal ions such as potassium and sodium (but not lithium), analogous to pure DNA and RNA quadruplexes. Moreover, the hybrid quadruplexes are very stable and readily form at low nanomolar concentrations (or even lower). Thus, a sequence-targeted approach to recognizing these important regulatory regions of DNA or RNA has the potential to create high affinity, specific regulators of gene expression.

We recently compared complementary and homologous hybridization of PNA probes to an RNA quadruplex and found that 7 base PNAs were able to disrupt the folded RNA structure to bind to their target sequence (29). While the complementary PNA formed the expected 1:1 duplex structure with the RNA target, the homologous PNA surprisingly formed a 2:1 (i.e. PNA₂-RNA) hybrid quadruplex. This paper describes using the same probes to target a simplified RNA quadruplex, allowing thermodynamic data to be obtained for the hybridization reactions. In addition, homologous PNA probes were synthesized with both possible strand polarities (N-C versus C-N) and used to explore the issue of PNA strand alignment within hybrid quadruplexes. These experiments provide an increasingly detailed picture of hybrid PNA-RNA quadruplex formation and point the way toward further increasing the affinity of short G-rich PNAs for homologous targets in biological RNA.

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EXPERIMENTAL PROCEDURES

Chemicals. 2'-O-Ortho-ester-protected RNA oligonucleotides were purchased from Dharmacon Research (www.dharmacon.com). The RNA oligonucleotides were deprotected using TEMED—acetate buffer (pH 3.8) according to guidelines provided by the manufacturer. Stock solutions of the RNA oligonucleotide were prepared in water. Concentration determinations were based on the absorption at 260 nm and were measured at 80 °C on a Cary 3 Bio spectrophotometer. At 80 °C, the nucleobases are assumed to be unstacked and the molar absorptivity is then the sum of individual absorptivities of the RNA monomers. Values for the extinction coefficients for RNA were obtained from the literature (31).

t-Boc-protected PNA monomers were purchased from Applied Biosystems (www.appliedbiosystems.com) and were used to synthesize PNA probes by a solid-phase peptide synthesis method (32, 33). A PNA probe was labeled at the N terminus with a carboxylic acid derivative of the thiazole orange (TO) dye prior to cleavage, following the same procedure as for the coupling of PNA monomers. The synthesis of the thiazole orange derivative is described elsewhere (34, 35). PNA probes were purified via HPLC and characterized by mass spectrometry. Mass spectra were recorded on a matrix-assisted laser desorption ionizationtime-of-flight instrument (MALDI-TOF) (Applied Biosystems Voyager DE sSTR) using α-cyano-4-hydroxycinnamic acid as the matrix (P7C: m/z calculated for $[M + H]^+$, 1970.9; found, 1969.5. P7C_{TO}: m/z calculated, 2357.9; found, 2355.6. P7H: m/z calculated, 2121.98; found, 2119.78. P7H_{TO}: m/z calculated, 2508.28; found, 2505.50. P7Hr: m/z calculated, 2121.98; found, 2120.68). Extinction coefficients for PNA monomers were obtained from Applied Biosystems $(C = 6600 \text{ M}^{-1} \text{ cm}^{-1}, T = 8600 \text{ M}^{-1} \text{ cm}^{-1}, A = 13,700)$ M^{-1} cm⁻¹, and $G = 11700 M^{-1}$ cm⁻¹). All experiments were carried out in buffer containing 10 mM Tris-HCl (pH 7.0) and various amounts of KCl or LiCl as denoted in the text or figure captions.

UV Melting Curves. UV—vis measurements were performed on a Varian Cary 3 spectrophotometer equipped with a thermoelectrically controlled multicell holder. RNA—PNA hybrids were prepared by mixing equimolar amounts of the single strands in buffer, heating to 95 °C, equilibrating for 5 min, and then cooling (annealing) to 15 °C at a rate of 1 °C/min while monitoring the change in absorbance signal at 275 and 295 nm with a data interval of 0.5 °C. Subsequent reheating under the same condition revealed no hysteresis effect

Circular Dichroism (CD). CD experiments were performed on a Jasco J-715 spectrophotometer. All spectra represent an average of six scans and were collected at a rate of 100 nm/min. All spectra were baseline-subtracted and smoothed via a nine-point adjacent averaging algorithm. Hybrids were prepared by heating to 95 °C followed by slow cooling to 20 °C prior to recording spectra. Temperature-dependent spectra were obtained by annealing to the lowest temperature of interest. The samples were equilibrated for 10 min at that temperature and then scanned (as above). Samples were then heated by 10 °C at a rate of 1 °C/min and equilibrated at that temperature for 5 min before subsequent scanning.

Fluorescence Spectroscopy. Fluorescence emission spectra were collected on a Photon Technology International (PTI) fluorescence system. The emission spectra of RNA + PNA $_{\rm TO}$ samples were measured upon excitation of TO at 495 nm. The spectral bandwidths used were either 4 or 5 nm. Samples containing appropriate amounts of PNA and RNA were prepared in buffer containing 10 mM Tris-HCl (pH 7.0) and 100 mM KCl and annealed prior to recording the spectra.

Determination of Binding Constants. Stock solutions (10 μ M) of RNA targets were prepared in buffer containing 10 mM Tris-HCl (pH 7.0) and 100 mM KCl and annealed. RNA targets were titrated into 0.1 μ M solutions containing labeled PNAs (either P7H_{TO} or P7C_{TO}) prepared in the same buffer. Fluorescence spectra recorded 10 min after each addition of RNA exhibited increasing intensity, indicative of the hybridization of PNAs to the RNA targets. On the basis of the concentration dependence of the fluorescence and assuming 100% binding at the saturation of fluorescence signal, the fitting to a 1:1 binding model (eq 1) yielded the equilibrium binding constants (36)

$$\frac{F}{F_{\infty}} = \frac{F_0}{F_{\infty}} + \frac{F_{\infty} - F_0}{2[P]_0} \frac{1}{F_{\infty}} (b - \sqrt{b^2 - 4[P]_0[G]_0})$$
where $b = [P]_0 + [G]_0 + K_D$ (1)

 $[P]_0$ represents the concentration of PNA used, $[G]_0$ is the concentration of RNA at each point in the titration, and F_0 , F, and F_∞ are the fluorescence intensities of PNA in solution, PNA + RNA at each point of the titration, and final value of fluorescence at the saturation point, respectively. The "fraction bound" was calculated by dividing the fluorescence at each point in the titration (F) by the fluorescence at the saturation point (F_∞) . The binding constant was calculated as $1/K_D$.

Determination of Binding Stoichiometry. Continuous variation experiments (37) for the RNA–PNA_{TO} binding were performed by mixing variable ratios of TO-labeled PNA and RNA targets together but keeping the total concentration constant (0.5 μ M). Samples were prepared in buffer containing 10 mM Tris-HCl (pH 7.0) and 100 mM KCl and were annealed prior to recording fluorescence spectra as described before. Emission at 535 nm was plotted against the mole fraction of PNA_{TO}, and stoichiometry was determined from the maximum point in the plot.

RESULTS

RNA Target Characterization. RDQ was originally selected from a combinatorial RNA library for binding to the fragile X mental retardation protein (12). While it is not a natural target, it represented a useful RNA quadruplex model system. In our previous paper, we targeted RDQ with complementary and homologous PNAs P7C and P7H (Chart 1). The RNA consists of duplex and quadruplex domains based on secondary-structure prediction and ¹H NMR experiments (14). The presence of the stem region in the RNA complicates analysis of PNA hybridization because of its background absorbance. Therefore, we designed a truncated version in which the stem is deleted (RQ). The extra nucleotides separating the duplex and quadruplex domains in RDQ were retained in RQ because of the fact that

Chart 1: RNA Targets, PNA Probes, and Thiazole Orange Dye Structure^a

RNA Targets

PNA Probes P7C H-CCCAGCC-Lys-NH₂ Р7Сто TO-CCCAGCC-Lys-NH2 H-GGGTCGG-Lys-NH2 P7H_{TO} TO-GGGTCGG-Lys-NH₂ H-GGCTGGG-Lys-NH2 P7Hr RQ Thiazole Orange (TO) PNA Base

^a Target sequence for PNAs is shown in blue.

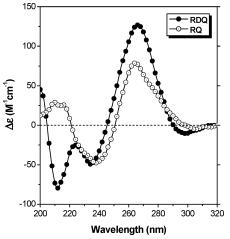


FIGURE 1: CD spectra recorded at 20 °C for RDQ and RQ. [RNA] = 2.0 μ M, and [KCl] = 100 mM.

overhanging nucleotides have been shown to stabilize PNA-DNA (27) and PNA-RNA (38) hybrids.

Figure 1 shows CD spectra recorded for the two RNAs in aqueous buffer. The spectra have similar shapes in the region of 225-300 nm but differ at 210 nm. The negative peak evident in the spectrum for RDQ is characteristic of duplex RNA (39); therefore, it is not surprising that this feature is lost when the stem is deleted to give RQ.

Meanwhile, the UV melting curves shown in Figure 2 illustrate how deletion of the stem simplifies thermal analysis of the RNA. The full-length RNA exhibits two transitions when the absorbance is monitored at 295 nm. First, a hypochromic transition is observed with a melting temperature $(T_{\rm m})=$ ca. 45 °C. Further heating leads to a hyperchromic transition with $T_{\rm m}={\rm ca.~80~^{\circ}C.}$ These transitions

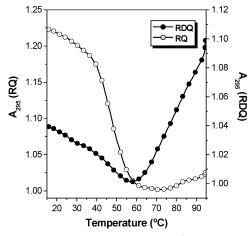


FIGURE 2: UV melting curves recorded at $\lambda = 295$ nm for RDQ and RQ. [RNA] = $2.0 \mu M$, and [KCl] = 100 mM. Data were collected every 0.5 °C during a heating ramp of 1 °C/min.

have been assigned to quadruplex melting at lower temperature to give a large hairpin structure followed by melting of the stem to produce the open, random-coil RNA (29). This interpretation is supported by prior work showing that, in DNA, melting of G quadruplexes yields hypochromic transitions when monitored at 295 nm (40), while Watson-Crick duplexes exhibit hyperchromic transitions. Moreover, using LiCl instead of KCl in the buffer reduces the lower temperature transition in RDO without affecting the higher temperature transition (29). This is consistent with the known preference of G quadruplexes for K⁺ over Li⁺.

Deletion of the stem results in a much simpler melting curve for RQ (Figure 2). A single hypochromic transition is observed at approximately the same temperature where the quadruplex domain in RDQ melts. The relatively clean

Table 1: Melting Temperatures and Thermodynamic Parameters^a for Folding of RQ and Hybridization with Complementary (P7C) and Homologous (P7H) PNAs

	$-\Delta G_{298 \text{ K}}$ (kcal/mol)	$-\Delta H$ (kcal/mol)	$-T\Delta S_{298 \text{ K}}$ (kcal/mol)	T _m (°C)
RQ	2.7 ± 0.5	39.5 ± 1.1	36.8 ± 1.0	47.2 ± 1.0
RQ + P7C	12.3 ± 0.4	43.0 ± 0.4	30.6 ± 2.0	58.7 ± 0.8
RQ + 2P7H	28.5 ± 0.4	95.7 ± 1.9	67.1 ± 2.0	70.1 ± 0.5

^a Determined by curve fitting (41).

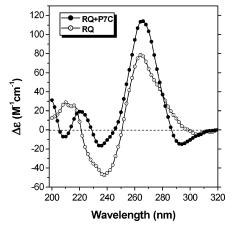


FIGURE 3: CD spectra recorded at 20 °C for RQ alone and with equimolar P7C. [RQ] = $2.0~\mu\text{M}$, and [KCl] = 100~mM.

transition allows thermodynamic parameters for quadruplex melting to be extracted from the curve (Table 1). Replacing KCl with LiCl in the buffer decreases the melting transition by ca. 20 °C (Figure S1 in the Supporting Information), supporting the assignment of this transition to melting of a G quadruplex.

While the topology of the G quadruplex in RQ cannot be determined solely by the CD spectral shape and melting curve, the results presented above demonstrate that the RNA folds into a stable secondary structure based on stacked G quartets and should serve as a suitable target for hybridization by complementary and homologous PNA probes.

Hybridization of Complementary PNA to RQ. The 7 base PNA P7C is complementary to G13–G19 within RQ (numbering from the 5' end and shown in blue in Chart 1). Hybridization of the PNA requires disruption of the quadruplex structure to form PNA–RNA base pairs with G13, G14, G18, and G19. Mixing P7C with RQ in equal concentrations yields the CD spectrum shown in Figure 3. The shape of the spectrum is similar to that of RQ alone, except at 210 nm. PNA–RNA duplexes exhibit CD maxima at 265 nm and minima at 240 nm (42), consistent with the spectrum shown in Figure 3.

Hybridization between P7C and RQ was also studied using a fluorescent-labeled version of the PNA. Thiazole orange (TO) was linked to the PNA N terminus to serve as a fluorescent reporter for hybridization (35). Upon hybridization with RQ, the TO dye should either stack on the end of the PNA-RNA hybrid or bind between unpaired bases on the RNA strand that extend beyond the PNA-binding site. In either case, the conformational freedom of the dye will be reduced, leading to enhanced fluorescence (43, 44). This behavior is not unique to PNA-linked TO. In fact, unsymmetrical cyanine dyes are well-known for their enhanced

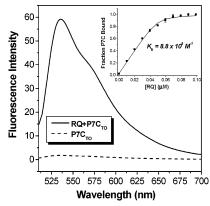


FIGURE 4: Fluorescence spectra recorded for P7C_{TO} alone or with equimolar RQ. [PNA] = $2.0~\mu$ M, and [KCl] = 100~mM. Samples were excited at 495 nm. (Inset) Binding isotherm obtained from titrating RQ into P7C_{TO} with a fit to the 1:1 binding model.

fluorescence quantum yields when bound to DNA or RNA (45-47).

Figure 4 illustrates fluorescence emission spectra recorded for P7C_{TO} alone in solution or in the presence of RQ. The fluorescence increases 27-fold in the presence of the RNA target, consistent with successful hybridization of the PNA. A continuous variations experiment verified that a 1:1 complex was formed (Figure S2 in the Supporting Information) between the PNA and RNA. An experiment in which RQ was titrated into a solution of P7C_{TO} was then performed with the fluorescence intensity recorded after each addition. The data were used to generate the binding isotherm shown in the inset to Figure 4, where fitting to a 1:1 binding model yielded an equilibrium constant (K_b) of 8.8 × 10⁸ M⁻¹ cm⁻¹. The same experiment performed using RDQ as the target yielded $K_b = 8.1 \times 10^8 \text{ M}^{-1} \text{ cm}^{-1}$ (29), indicating that the stem of RDQ has almost no effect on the affinity of the PNA probe for the quadruplex target. These results illustrate that a short PNA oligomer can disrupt an RNA quadruplex secondary structure and form a PNA-RNA duplex at low nanomolar concentrations.

UV melting curves were recorded at two wavelengths (275 and 295 nm) to study PNA-RNA hybridization. Figure 5A shows that the hypochromic transition observed at 295 nm and assigned above to the melting of the RQ quadruplex is lost in the presence of P7C, verifying that hybridization of the PNA disrupts the RNA secondary structure. Meanwhile, a hyperchromic transition is observed at 275 nm (Figure 5B), and this is assigned to denaturation of the PNA-RNA hybrid duplex. Because the melting temperature for the duplex is higher than that of the quadruplex, dissociation yields the unstructured RNA rather than the refolding of RQ into its quadruplex structure. Thermodynamic parameters for PNA-RNA hybridization were determined from the melting curve at 275 nm and are shown in Table 1. The equilibrium constant (K_b) calculated from the free-energy change using the formula $\Delta G = -RT \ln K_b$ is $1.0 \times 10^9 \text{ M}^{-1}$, which matches well with that determined by fluorescence titration (Figure 4).

Hybridization of Homologous PNA to RQ. PNA P7H is homologous to the same region of RQ targeted by the complementary PNA P7C, i.e., G13—G19. With a sufficient driving force, the PNA should be able to disrupt the RNA quadruplex and form a hybrid PNA—RNA quadruplex. In

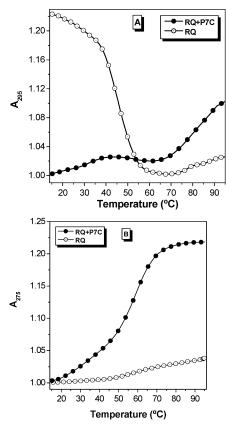


FIGURE 5: UV melting curves recorded at 295 nm (A) and 275 nm (B) for RQ alone or with equimolar P7C. [RNA] = $2.0 \mu M$, and [KCl] = 100 mM. Data were collected every $0.5 \,^{\circ}\text{C}$ during a heating ramp of $1 \,^{\circ}\text{C/min}$.

fact, prior work showed that P7H bound to RDQ in a 2:1 stoichiometry, meaning two PNAs associated with a single RNA (29). We speculated that both PNAs were forming quadruplexes, one with each "half" of the original RNA quadruplex. We sought to verify this using the simplified construct RQ.

Before performing hybridization experiments, we characterized the behavior of P7H in the absence of RNA. Guaninerich PNAs have been shown to form homoguadruplex structures (48, 49), and the extent to which this would interfere with P7H hybridization to RQ was unknown. The behavior of P7H in aqueous buffer was studied by CD and UV melting curves. A weak CD spectrum was recorded for the PNA, but a relatively high concentration of the PNA (100 μM) was required (Figure S3A in the Supporting Information). At this concentration, a clear hypochromic transition was observed in the UV melting curve recorded at 295 nm (Figure S3B in the Supporting Information), indicating that P7H does form a homoquadruplex. However, at the lower concentrations used for the hybridization experiments (<5 μ M), P7H exhibits neither a CD spectrum nor a UV melting transition. Therefore, we consider the PNA to be monomeric under the conditions of the following experiments.

Figure 6 illustrates fluorescence spectra recorded for the TO-labeled analogue P7H_{TO} alone and in the presence of RQ. The fluorescence enhancement is only 2.8-fold, compared with 27-fold for the complementary PNA targeted to the same RNA. The large difference in enhancement for the two probes is due to the background fluorescence of the unhybridized probe: the fluorescence is more than 10-fold

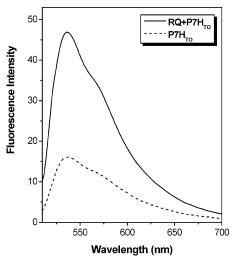


FIGURE 6: Fluorescence spectra recorded for P7H $_{TO}$ alone or with equimolar RQ. [PNA] = 2.0 μ M, and [KCl] = 100 mM. Samples were excited at 495 nm.

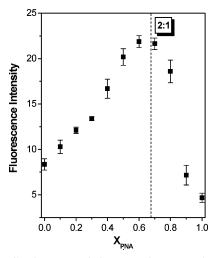


FIGURE 7: Continuous variation experiment to determine the stoichiometry of the $P7H_{TO}-RQ$ hybrid. Fluorescence intensity was recorded at 536 nm.

higher for P7H_{TO} than for P7C_{TO}. The free-probe fluorescence of TO-conjugated PNAs is highly dependent upon the sequence and has been attributed to the binding of the fluorophore within a folded or collapsed conformation of the PNA (50). The purine-rich P7H_{TO} should be more hydrophobic than the pyrimidine-rich P7C_{TO}; therefore, it is not surprising that the G-rich PNA exhibits higher intrinsic fluorescence. A continuous variation experiment indicates the formation of a 2:1 PNA–RNA hybrid (Figure 7), analogous to what was found previously with RDQ (29).

UV melting curves were used to further characterize the P7H-RQ hybrid. Figure 8 compares melting curves recorded at 295 nm for RQ alone and with 1 or 2 equiv of the homologous PNA. In contrast to the complementary PNA, hybridization of RQ with P7H yields a complex that melts with a hypochromic transition (compare with Figure 5A). The transitions for the two PNA-containing samples occur at substantially higher temperatures than for RQ alone, and the $T_{\rm m}$ is slightly higher when 2 equiv of PNA is used. More striking is the variation in hypochromicity for the samples. The change in absorbance for RQ alone and with 1 equiv of PNA is comparable, but adding a 2nd equiv of PNA increases

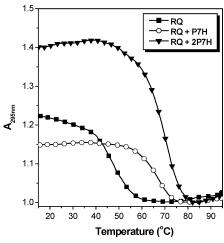


FIGURE 8: UV melting curves recorded at 295 nm for RQ alone or with 1 or 2 equiv of P7H. [RNA] = 2.0 μ M, and [KCl] = 100 mM.

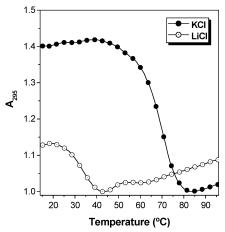
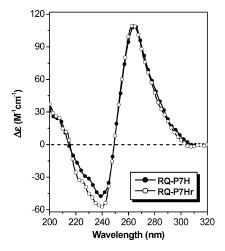


FIGURE 9: UV melting curves recorded at 295 nm for the RQ-2P7H hybrid. [RNA] = $2.0 \mu M$, and [KCI] = [LiCl] = 100 mM.

the hypochromicity by a factor of 2-3. Finally, substituting KCl with LiCl in the buffer decreases the $T_{\rm m}$ of the RQ-2P7H hybrid by ca. 40 °C (Figure 9), verifying that the complex consists of a hybrid quadruplex.

Thermodynamic parameters for hybridization of P7H with RQ were determined from the 2:1 UV melting curves shown above and are listed in Table 1. The hybrid quadruplex is much more stable than the RQ target ($\Delta G_{298~K} = -28.5$ and -2.7 kcal/mol, respectively), explaining why the PNA easily disrupts the intramolecular quadruplex structure of RQ.

PNA Strand Orientation. While complementary hybridization of PNA with RNA is more favorable when the N terminus of the PNA aligns with the 3' terminus rather than the 5' terminus of the RNA (42), orientational preferences for hybridization of homologous PNA and RNA strands are unknown. Therefore, we synthesized the "reverse" homologous PNA in which the sequence is inverted to give P7Hr (Chart 1). This PNA hybridized with RQ to give CD spectra and UV melting curves that were nearly identical to those obtained with P7H-RQ (Figure 10), indicating that there is no orientational preference for hybrid quadruplex formation between RQ and short, homologous PNA oligomers. (Interestingly, the complementary PNA-RNA duplex and homologous PNA-RNA quadruplex give similar CD spectra. However, the opposite melting curve profiles monitored a



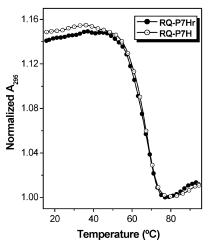


FIGURE 10: CD spectra (top) and UV melting curves recorded at 295 nm (bottom) for the RQ-P7H and RQ-P7Hr hybrid quadruplexes. [RNA] = [PNA] = $2.0~\mu$ M, and [KCl] = 100~mM.

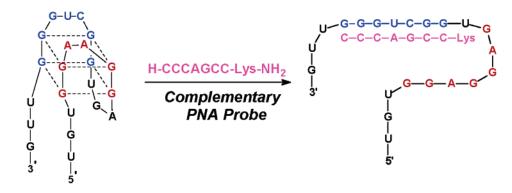
295 nm as well as the dependence on cation identity to distinguish between the two types of hybrids.)

DISCUSSION

The truncated RNA oligonucleotide RQ was used as a model for the full-length RDQ that we studied previously because the deletion of the stem nucleotides was expected to simplify the UV melting curves. This was verified, as shown in Figure 2, where a single hypochromic transition corresponding to quadruplex melting is observed for RQ, compared with two transitions for RDQ. In addition to eliminating the high-temperature stem-melting transition, the deletion of the stem results in a ca. 3-fold larger hypochromicity than in the full-length RNA, even though the melting temperatures are equivalent. The improvement in signal strength for RQ is most likely due to the deletion of 15 nucleotides from RDQ that absorb light at 295 nm but do not contribute to the hypochromicity.

One other aspect of the RNA targets is worthy of comment. The secondary structures shown in Chart 1 are based on previous literature papers for RDQ (12, 14) and on the similarity of RQ to RDQ. The topologies of these structures are reminiscent of various DNA quadruplexes, including the thrombin-binding aptamer, in which adjacent G tracts have antiparallel orientations (19, 20). The CD spectrum for that DNA exhibits a maximum at 295 nm (51).

Complementary Hybridization



Homologous Hybridization

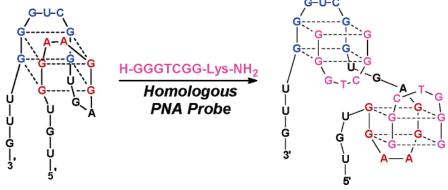


FIGURE 11: Comparison of complementary and homologous hybridization modes for PNA-RNA recognition. Note that there are several possible structures for the 2:1 hybrid quadruplex in addition to the one shown above.

In contrast, the CD spectra for both RDQ and RQ exhibit maxima at 265 nm and minima at 240 nm. These features are similar to those exhibited by other G quadruplexes, such as that formed by a DNA model sequence based on a transcriptional regulatory region for the c-MYC oncogene (52). NMR spectroscopy has demonstrated that in the c-MYC quadruplex, adjacent G tracts have parallel orientations, with the chain folding back twice on itself between successive G tracts via the loop nucleotides (i.e., the "double-chain reversal") (52, 53). In another study, Oliver and Kneale reported that a DNA oligomer expected to form a parallel four-stranded quadruplex and its RNA homologue gave similar CD spectra, suggesting that the RNA formed a parallel structure as well (54). Thus, while CD spectra alone are not sufficient to determine strand orientations (55-57), these observations raise the possibility that RQ and RDQ adopt parallel rather than antiparallel folds.

Complementary versus Homologous Hybridization. Hybridization of the complementary PNA P7C to RQ yields a duplex structure with a CD spectrum similar to that reported previously for a 15-bp PNA-RNA hybrid (42). The melting temperature ($T_{\rm m} = 58.7~{\rm ^{\circ}C}$) and free-energy change ($\Delta G_{\rm 298~K}$ = -12.3 kcal/mol) reflect not only the inherent high affinity of PNA for RNA but also the stabilizing effect of overhanging nucleotides as well as the high G-C content of the duplex. In prior work, we hybridized a similar 7 base PNA (H-CCACACC-LysNH₂) to the central 7 nucleotides of the quadruplex-forming DNA 5'-GGTTGGTGTGGTGG-3' and obtained similar hybrid stability: $T_{\rm m} = 55.2~{\rm ^{\circ}C}$ and $\Delta G_{\rm 298~K}$

= -11.7 kcal/mol (27), although those measurements were performed at lower KCl concentrations (10 versus 100 mM). The top part of Figure 11 depicts the formation of the PNA-RNA hybrid duplex at the expense of the RNA quadruplex secondary structure.

Meanwhile, the homologous PNA P7H hybridizes to RQ to form a ternary PNA2-RNA quadruplex structure. This suggests that the PNA invades the RNA quadruplex and then binds to the two halves of the target. One possible structure is shown in the lower part of Figure 11, although the RNA quadruplex could also unfold by disrupting the hydrogen bonds between the two blue G tracts and those between the two red G tracts.

The thermodynamic parameters determined from the melting curve of the ternary complex are shown in Table 1 and indicate that the formation of the 2:1 hybrid quadruplex is much more favorable enthalpically than the formation of the 1:1 hybrid duplex when the complementary PNA P7C is used. This finding is due at least in part to the larger number of hydrogen bonds involved in the formation of the 2:1 quadruplex (32 hydrogen bonds) versus the duplex (20 hydrogen bonds), although differences in base stacking, hydration, and coordination of potassium ions by the G tetrads likely contribute as well. Similarly, the additional loss of translational entropy incurred upon the formation of a ternary complex compared to a binary complex accounts for part of the more negative ΔS for the quadruplex. Conformational entropy is also an important consideration but more difficult to assess. While the formation of the 1:1

FIGURE 12: Four possible 1:1 hybrid quadruplexes formed by homologous RNA and PNA strands. In structures A and B, the PNA N terminus aligns with the RNA 5' terminus, whereas the orientation is reversed for structures C and D.

duplex by P7C leaves unpaired overhangs of 3 and 11 nucleotides on the 3' and 5' termini, respectively, the 2:1 quadruplex formed by P7H leaves as many as 14 nucleotides unpaired on the RNA strand and 6 nucleotides on the two PNA strands (assuming no interaction between the loop bases). Regardless of how the contributions to the ΔH and ΔS are parsed, the hybrid quadruplex formed by P7H with RQ is very stable.

Hybrid quadruplex formation between homologous PNA and RNA strands is more complicated than hybrid duplex formation between complementary strands because there are at least four possible structures that could form (Figure 12). These vary not only in terms of strand alignment, i.e., whether the PNA N or C terminus aligns with the RNA 5' terminus, but also with respect to the placements of the loops in the hybrid. We were able to assign strand orientations in the PNA₂-DNA₂ four-stranded quadruplex formed from the G₄T₄G₄ sequence by a set of fluorescence resonance energytransfer experiments (30). Analogous experiments in the present system would be less informative because of the compact structure of both the target RO and the hybrid quadruplex. However, we find that the CD spectrum of the hybrid PNA2-RNA quadruplex has a similar shape to the PNA₂-DNA₂ quadruplex, suggesting that the strand orientations are similar for the two hybrids. This implies alignment of the PNA N terminus with the RNA 5' terminus, eliminating the last two structures shown in Figure 12. Whether the RNA and PNA loops are on the same or opposite ends of the stacked G quartets cannot be determined at this time but will be the subject of future investigation. Note however that there is unlikely to be a strong interaction between the loops, because inverting the orientation of the PNA (P7Hr versus P7H) has no discernible effect on the PNA₂-RNA CD spectrum or melting temperature.

Further Enhancements in Affinity. The results presented above illustrate that both complementary and homologous PNAs can hybridize with high affinity to G-quadruplexforming RNA targets. Further improvement in the affinity of complementary PNAs is straightforward and primarily involves lengthening the PNA so that additional base pairs can form. The homologous PNAs require a different approach if higher affinity is needed. One possibility is to covalently attach two homologous PNAs to reduce the translational entropic penalty of hybridization (58, 59). Two considerations in designing such a bisPNA are (1) the positions at which the two PNA strands should be connected and (2) preventing the bisPNA from folding into its own stable quadruplex. Structural analysis of a PNA₂-RNA or PNA₂-DNA quadruplex would be most helpful in addressing the first issue. The very low stability of the homoquadruplex formed by P7H suggests that, even if a bimolecular quadruplex were to be formed by a bisPNA analogue, it should not significantly interfere with hybridization. Separately, quadruplex-binding small molecules could be attached to the PNA to "cap" the G-quartet stack, further stabilizing the hybrid quadruplex structure.

Finally, the ability of two homologous PNA strands to bind to RQ requires reconsideration of just what is meant by "homologous". In the strictest definition, two homologous sequences are identical at every position. However, this is clearly unnecessary for P7H to bind to RQ because there is only one 7 nucleotide sequence within the RNA that is completely homologous to the PNA, and yet, two PNAs hybridize to the RNA. Rather, it seems that homology in the G tracts is sufficient to permit hybrid quadruplex formation: RQ has four G2 tracts allowing two PNAs, each with two G₂ tracts, to simultaneously hybridize. If there is no direct interaction between the loop nucleobases (i.e., those not involved in G-quartet formation), then the sequence selectivity of hybridization could suffer, because there are likely to be many potential binding sites consisting of two G₂ tracts separated by a few nucleotides. Future experiments will focus on this issue by varying the number of nucleobases separating two G tracts within a given target and assessing hybrid quadruplex formation by G-rich PNAs.

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SUPPORTING INFORMATION AVAILABLE

Figure S1, UV melting curves recorded at 295 nm for RQ with either 100 mM KCl (●) or LiCl (○) included in the buffer; Figure S2, continuous variations experiment to determine stoichiometry of the P7CTO−RQ hybrid; and Figure S3, characterization of P7H in 10 mM Tris-HCl buffer (pH 7.0) and 100 mM KCl. This material is available free of charge via the Internet at http://pubs.acs.org.

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